CHEM 114L:
General Chemistry II lab
South Dakota School of Mines and Technology
Spring 2013
1 Credit Hour

Contact Information
Section 2 and 4 Instructor: Justin Meyer
Office: Chemistry Building 122
Phone: 605-394-2431
E-mail: Justin.Meyer@sdsmt.edu
Office Hours: MWF 9:00 AM to 10:30 AM

Section 1 and 3 Instructor: Tsventanka Filipova
Office: Chemistry Building 120
Phone: 605-394-1698
E-mail: Tsvetanka.Filipova@sdsmt.edu
Office Hours: Wed/Fri 1-2 pm; Thursday 10:15 – 12:30 pm

Section Meeting Times
Chem 114L M051 Monday 11:00 AM to 1:50 PM
Chem 114L M052 Tuesday 8:00 AM to 10:50 AM
Chem 114L M053 Thursday 2:00 PM to 4:50 PM
Chem 114L M054 Tuesday 12:00 Noon to 2:50 PM

CBEC 114

Catalog Description: Laboratory designed to accompany CHEM 114.

Course Prerequisites: Prerequisite: Chem112L. Prerequisite or co-requisite: Chem114. Lab is designed to accompany Chem114. If you have not completed Chem114 and are currently enrolled in Chem114, dropping the lecture may require you to drop the lab as well, since Chem114 is a co- or prerequisite.

Instructional Methods: Pre-laboratory discussion and "hands-on" experiment

REQUIRED TEXT AND EQUIPMENT:
1. Lab manual from Thomson Custom Solutions (ISBN- 10: 0-495-40783-6). A complete set consists of the following numbered experiments: 363, 364, 365, 366, 458 and 504. These books can be found at the bookstore. Additional labs (Estimating the Calorie Content of Foods, etc.) will be provided for you through D2L.
2. Approved safety goggles, which must be worn at all times while in the laboratory.
   Goggles may be purchased in the bookstore. Your goggles may be kept in your lab drawer.
3. Lab Notebook. A hardbound notebook should be purchased for use as a lab notebook. You will be informed of how and when to start using this notebook for experiments 364, 365 and 366.
5. Computer or Tablet. You should have access to a tablet or computer to complete assignments for the class. This may be outside of lab.
6. Blue or black non-erasable pen. Water resistant recommended. Points will be lost for using pencil or ink colors not included here.
COURSE POLICIES:

Attendance: Attendance at pre lab lectures and labs is mandatory. Tardiness may result in not being allowed to attend.

Make-up policy: Lab space is limited and you should not assume that you can attend another lab section other than the one in which you are officially registered. For those that have an excused absence, as determined by the instructor, a makeup lab will be given at the end of the semester. You will be given instructions on the makeup lab as needed.

Please note that the make-up lab is granted only for those students who have obtained an excused absence from the lab instructor. A legitimate reason, as determined by the instructor, includes school sponsored activities, illness requiring medical treatment, death in immediate family, or military obligations. In any case, the student needs to bring the appropriate documentation to the instructor prior to the scheduled lab time, or as soon as possible after. Excused absences for illness will require medical documentation.

The grade given on the make-up lab will replace the zero for the lab absence. The make up lab is scheduled for the last day that your lab section meets. If you miss the make up lab, scores for any excused absences will remain zero. There is no make up for missing the make up lab and check out meeting.

If a student misses the known part of one of the three qualitative labs, they will miss the 100 points for that part of the lab. If they miss the unknown part, they will miss the 80 points from that part of the lab. These can be made up in accordance with the policy outlined here. Missing EITHER the known or the unknown will count as an absence. Missing both will count as TWO absences.

*There is only one (1) make up lab. Students missing more than one lab for any reason will likely receive a score of zero (0) for the extra missed labs. Students missing three labs or more are not likely to pass the course. Each lab is about 10% of your grade, which is about a letter grade. Exceptional circumstances can lead to exceptions, but do not EXPECT to be an exception.*

Assessments/Grading: Final grades are determined based on the total points earned out of the 1050 total points in the course.

<table>
<thead>
<tr>
<th>Component</th>
<th>Points</th>
</tr>
</thead>
<tbody>
<tr>
<td>Lab scores (7 labs)</td>
<td>700 pts</td>
</tr>
<tr>
<td>Unknown Labs book</td>
<td>90 pts</td>
</tr>
<tr>
<td>Expt 363</td>
<td>10 pts</td>
</tr>
<tr>
<td>Unknown Report</td>
<td>150 pts</td>
</tr>
<tr>
<td>Discretionary</td>
<td>100 pts</td>
</tr>
<tr>
<td><strong>Total Pts:</strong></td>
<td><strong>1050 pts</strong></td>
</tr>
</tbody>
</table>

Once your cumulative total has been calculated, grades are assigned according to the following scale:

A $\geq$ 93.0%  B $\geq$ 83.0%  C $\geq$ 73.0%  D $\geq$ 63.0%  F < 63.0%

These levels MAY change, but they will not increase.

Lab Scores: Each lab is graded on a 100-point basis, except for Expt 363 which is worth a total of 10 points. The unknown lab scores are 80 points between the lab book and the report.
Some points to help with your pre and post labs:

- Be clear and concise when answering pre and post lab questions. Often a couple words will suffice; long drawn out explanations can lose more points than short concise answers. One can be too short, however. Make sure that you are clearly conveying a thought and not just a word or two.
- Long conclusions are often unnecessary, and can lose points for not being concise.
- Show your calculations and formulas used when you have to give a value. Credit may not be given for answers without clear indication of how they were arrived at. Units are always required.
- Chemical equations are often a really good way to help answer question, so you should include a supporting equation when needed. This will be expected as part of the answer for many problems.
- Double check and make sure to answer all questions, all parts of each question and answer what the question asks.
- As you are performing an experiment make sure you are aware not only of what you are doing but why. This will help keep you from getting off track during the course of an experiment and also make you aware of possible errors.

Pre labs: Read each lab and have the pre-lab assignment completed prior to class. All pre labs are due before you can enter your lab session. You may have to use your lecture class textbook for some of these prelabs as well! Nobody will be allowed to do the lab without completion of the pre lab. You will receive a score of zero for a lab you miss for not having the pre lab done. Pre labs are individual assignments. Incomplete pre labs may result in no credit for the lab when the grading is done. Make sure you finish the pre lab. No make-up lab will be allowed for missing a lab for not having the correct prelab DONE.

* If the instructor deems you have not prepared yourself properly for the lab you are about to do, they may ask you to leave and you will receive no credit for that period’s lab.*

Post labs: Post labs are due at the end of the lab period unless otherwise noted. Any late submissions will not be graded. Turn them in before you leave. Post lab questions are to be done IN CLASS and not before, unless otherwise specified for that experiment.

Discretionary Points: There will also be 100 discretionary points in the final grade. These points can be lost for a variety of reasons. For example, but not limited to: failure to obey laboratory safety rules, poor lab practice, tardiness, improper disposal of waste, wasteful handling of chemicals, behavior that is distracting and unprofessional in the lab. Acts of academic dishonesty will automatically result in the loss of all discretionary points. You will lose all of your remaining points for failure to check out!

Final Unknown Report: At the conclusion of each of the Qualitative Analysis labs (ANAL 364-366) you will have to identify an unknown. This will be an individual project for which you will be given one lab period. You will be using your results from the previous work in your lab book and the manual for this. You may work with your lab partner, but you will have a different unknown so you will have your own conclusions. You may not work with others in the lab besides your lab partner, if assigned. After this project is completed you will need to prepare a short report (3-5 pages) on your results from all three unknown portions. The report is an individual assignment. You will then need to submit your lab book and a hard copy of your short report when due. Further details will be given before the start of the qualitative project.
Reports or lab books handed in later than the end of your last lab period will lose half the possible points. Any reports or lab books handed in later than noon of the following day will not be graded, and given a score of zero (0). Details on the grading of the report/lab books will be provided before the start of the project.

**Lab Drawers and Partners (If assigned):** All labs will be done with partners, if assigned unless otherwise noted. At the beginning of the semester you will be assigned a partner who you will work with throughout the duration of the semester, subject to reassignment.

*If you are having problems with your lab partner that are affecting your performance, please see your instructor.*

**Course Outcomes:**
- Perform procedures for the analytical separation and qualitative determination of selected cations in an aqueous solution.
- Understand the fundamental and operational principles upon which common methods of separation and purification of chemical substances are based.
- Identify sources of error in chemical experiments.
- Interpret experimental results and draw reasonable conclusions.
- Practice laboratory safety procedures.
- Anticipate, recognize, and respond to hazards of chemical materials and manipulations.
- Learn the importance of following correct laboratory procedures.
- Keep **legible** and **complete** experimental records.
- Collaborate with peers in obtaining and interpreting data.

**Course Objective:** Students will gain familiarity with the principles and techniques of inorganic qualitative analysis, chemical kinetics, and the synthesis of selected chemical compounds.

**ADDITIONAL LABORATORY RULES**
- Admittance to the laboratory will be denied if a student does not have department-approved safety goggles. Modification of goggles is prohibited and may result in dismissal for the lab period.
- Students wearing improper or incomplete attire will be asked to leave the laboratory. They may be permitted to return when items in safety violation are replaced with acceptable clothing. Students sent for proper clothing will receive a grade penalty for the lab they miss.
- An unauthorized experiment at any time will result in the immediate assignment of a final grade of “F” for the course.
- Dispose of laboratory materials in proper waste bottles that are located in the hood(s). Note labels on waste bottles. If you have any doubt about where to dispose of something, ask your instructor for assistance.
- Any student exhibiting habitual disregard for any safety policy will be asked to leave the laboratory.
- Laboratory drawer replacement items are free of charge only on the first scheduled meeting. After the initial meeting, you will be responsible for the replacement of broken or missing items. This includes, but is not limited to, items that are lost due to your failure to return them to your drawer or your failure to lock your drawer. Replacement items are to be paid for at the time of their acquisition.
- Unexpected events (emergencies, spills, accidents, etc.) must be brought to the immediate attention of your instructor. Do not leave the lab without informing the instructor of the event.
• The use of tablet/laptop PCs in the lab is prohibited. You may store them out of harm’s way during the lab, but you may not use them in the lab. Use of them outside of the lab is encouraged. Limited coat rack space is available in the lab room.
• Students must vacate the lab by the designated end of the lab time. Students who want to work past this time will not be allowed to do so because of safety and scheduling issues. There IS a lab after your lab.

**Tentative Schedule:**

<table>
<thead>
<tr>
<th>Day of the week</th>
<th>Lab activity</th>
</tr>
</thead>
<tbody>
<tr>
<td>M</td>
<td>T</td>
</tr>
<tr>
<td>1/14</td>
<td>1/22</td>
</tr>
<tr>
<td>4/22</td>
<td>1/29</td>
</tr>
<tr>
<td>1/28</td>
<td>2/5</td>
</tr>
<tr>
<td>2/4</td>
<td>2/12</td>
</tr>
<tr>
<td>2/11</td>
<td>2/19</td>
</tr>
<tr>
<td>2/25</td>
<td>2/26</td>
</tr>
<tr>
<td>3/11</td>
<td>3/12</td>
</tr>
<tr>
<td>3/18</td>
<td>3/19</td>
</tr>
<tr>
<td>4/15</td>
<td>4/16</td>
</tr>
<tr>
<td>4/22</td>
<td>4/23</td>
</tr>
</tbody>
</table>

Labs marked with * above will be found on D2L. You will need to print these out and bring them to class. Monday sections will do Calorimetric Analysis of Food on the same day that they check out, 4/22/12 due to Monday holidays.

Subject to change.

<table>
<thead>
<tr>
<th>Other important dates</th>
</tr>
</thead>
<tbody>
<tr>
<td>Last day to add/drop.</td>
</tr>
<tr>
<td>Martin Luther King Day, no class</td>
</tr>
<tr>
<td>President’s Day, no class</td>
</tr>
<tr>
<td>Spring Break</td>
</tr>
<tr>
<td>Midterm.</td>
</tr>
<tr>
<td>Easter Break</td>
</tr>
<tr>
<td>Last day to withdraw.</td>
</tr>
<tr>
<td>Final exam week</td>
</tr>
</tbody>
</table>

**Fine for Failing to Check in at the End of the Semester:** All students who have checked out a lab equipment drawer and lock are required to check the drawer and lock back in at the end of the semester or earlier if withdrawing from the course, and to replace missing or broken items. A fine of $30.00 is assessed for failure to check in and a fine of $30.00 is assessed if you lose your lock or fail to turn it in upon check-in. If circumstances force you to withdraw from the lab before the end of the semester, you should make arrangements with the Chemistry Lab Manager (Margaret Smallbrock, C 123) to check in your equipment drawer and lock to avoid the fines.

**Cheating and plagiarism policy:** Any cheating as defined by the student code of conduct will not be tolerated in this course. See [http://sdmines.sdsmt.edu/sdsmt/studentconduct/main](http://sdmines.sdsmt.edu/sdsmt/studentconduct/main) regarding the student code of conduct. Cases of cheating will be handled on a case to case basis as defined in the student code of conduct. Please note that according to “Policy Governing Academic Integrity” in the SDSM&T
Undergraduate Catalog, the instructor of record for this course has discretion of how acts of academic dishonesty are penalized, subject to the appeal process, and that "Penalties may range from requiring the student to repeat the work in question to failure in the course." (72-73).

Any student who “dry labs” an experiment will be dismissed from the course with a grade of “F.” Borrowing data from another student, pretending to complete one or more of the laboratory activities, making up data, or turning in copies of other students’ work are all examples of dry labbing. This list is not complete and if you have concerns, please discuss them with the lab instructor.

Possession of data from previous semesters or sections or other unauthorized outside sources will result in dismissal from the course with a grade of “F.” Copying pre lab questions from another student current or former also counts as use of an unauthorized source.

Electronic Devices Policy: Please turn off your cell phone before your section starts. No text messaging in class. No headphones. No other use of any other electronic device or computer media is allowed in the laboratory. Persons in violation of this policy will be dismissed from the lab for the day on the first violation.

ADA Statement: Students with special needs or requiring special accommodations should contact the instructor (Joseph Marshall, 394-1679) and/or the campus ADA coordinator, Jolie McCoy, at 394-1924 at the earliest opportunity.

Freedom in learning: Students are responsible for learning the content of any course of study in which they are enrolled. Under Board of Regents and University policy, student academic performance shall be evaluated solely on an academic basis and students should be free to take reasoned exception to the data or views offered in any course of study. Students who believe that an academic evaluation is unrelated to academic standards but is related instead to judgment of their personal opinion or conduct should contact the dean of the college which offers the class to initiate a review of the evaluation.

Syllabus is subject to change.